Acta Crystallographica Section C Crystal Structure Communications

ISSN 0108-2701

L-Alanylglycyl-L-alanine monohydrate at 20 K

Diana Förster, Marc Messerschmidt and Peter Luger*

Institut für Chemie, Kristallographie, Freie Universität Berlin, Takustraße 6, 14195 Berlin, Germany

Correspondence e-mail: luger@chemie.fu-berlin.de

Received 11 April 2005 Accepted 10 May 2005 Online 11 June 2005

The X-ray crystal structure of the title compound, $C_8H_{15}N_3O_4$ ·H₂O, at 20 K (space group $P2_1$) reveals that the molecular conformation of the tripeptide is remarkably different from the water-free form (space group $P2_12_12_1$) reported previously [Padiyar & Seshadri (1996), *Acta Cryst.* C**52**, 1693–1695].

Comment

In the course of our ongoing charge-density investigations on oligopeptides, we became interested in some comparative studies of tripeptides of the type Ala-Xxx-Ala, where Xxx was to be varied among the 20 naturally encoded amino acids. An initial charge-density study was performed on the homotripeptide (L-Ala)₃ (Rödel, 2003). In a further step, we considered the system L-alanylglycyl-L-alanine, the conventional X-ray crystal structure of which (with spherical scattering factors) had previously been determined at room temperature (Padiyar & Seshadri, 1996). In a number of recrystallization attempts, we were unable to reproduce the literature modification. However, we were able to grow crystals of the title monohydrate, (I), while the structure reported by Padiyar & Seshadri (1996) is solvent free. We present here the previously unreported structure of L-Ala-Gly-L-Ala-H₂O, (I), based on a low-order X-ray data set (d = 0.70 Å) taken at 20 K.

$$\underset{M_{e}}{\overset{H_{3}}{\overset{H}{\underset{O}}}} H \underset{H_{H}}{\overset{H}{\underset{H}}} H \underset{H_{H}}{\overset{O}{\underset{H}}} H \underset{O}{\overset{H}{\underset{H}}} O^{-} H_{2}O$$

The structure of the asymmetric unit of (I), together with the atomic numbering scheme, is shown in Fig. 1, with the displacement parameters at 20 K. The tripeptide exists in the zwitterionic form in the crystal. Bond lengths and angles are as expected and need no detailed discussion. A tendency of the 20 K structure to have slightly longer bonds than the roomtemperature water-free structure (up to 0.027 Å longer) could be expected.

Apart from the space group $(P2_12_12_1)$ for the water-free form and $P2_1$ for the monohydrate), the two conformational

isomers differ considerably in the torsion angles of the terminal groups (Table 1 and Fig. 2). This holds for ψ_1 at the N terminus and is even more pronounced for φ_3 . This angle is -159.1 (2)° for (I), indicating an almost *trans* arrangement of the carboxylate group with respect to the N3–C4 peptide bond, while it is -71.3 (2)° for the water-free form, describing a *gauche* conformation in this region. All further conformational torsion angles of the two molecules agree to within 5–15°.

The crystal packing of (I) is illustrated in Fig. 3, with the hydrogen bonds indicated by dashed lines. The hydrogen bonds are also summarized in Table 2. The twofold screw axis generates a head-to-tail arrangement of two molecules in the *ac* plane linked by three hydrogen bonds, *viz*. N1–H11A···O4ⁱ, N3–H13···O1ⁱ and O4···H11Cⁱⁱⁱ–N1ⁱⁱⁱ (symmetry codes as in Table 2). The screw axis also generates a helical linkage of this molecular pair in the *b* direction. Molecules related by a pure translation in the *c* direction are connected by a bifurcated hydrogen bond from the ammonium N1–H11*B* to both oxygen acceptors of the carboxylic acid group, with the N1–H11B···O4ⁱⁱ hydrogen bond being rather weak. A further interaction in the *ac* plane is also seen in the *a* direction, where neighbouring layers are linked *via* the





The molecular structure and atom-numbering scheme of (I). Displacement ellipsoids are plotted at the 50% probability level and H atoms are shown as small spheres of arbitrary radii.



Figure 2

A fit of L-alanylglycyl-L-alanine in the monohydrate (this work) and water-free forms (Padiyar & Seshadri, 1996).



Figure 3

The crystal packing of (I), projected on to the *ac* plane. Hydrogen bonds are indicated by dashed lines and symmetry codes are as in Table 2.

water molecule, establishing a cycle of four (three independent) hydrogen bonds, viz. N2-H12···O5^{iv}-H25 A^{iv} ···O2-C4-C3-N2 (symmetry code as in Table 2). Except for two weak C-H···O interactions, no further intermolecular contacts of interest exist.

Experimental

Crystals of (I) were grown by evaporation from an aqueous solution of L-alanylglycyl-L-alanine (Bachem, Germany).

Crystal data

S = 1.07

1851 reflections

213 parameters

$C_8H_{15}N_3O_4 \cdot H_2O$	$D_x = 1.366 \text{ Mg m}^{-3}$
$M_r = 235.25$	Mo $K\alpha$ radiation
Monoclinic, P2 ₁	Cell parameters from 1009
a = 10.207 (6) Å	reflections
b = 4.780 (3) Å	$\theta = 2.9-24.7^{\circ}$
c = 11.955 (7) Å	$\mu = 0.11 \text{ mm}^{-1}$
$\beta = 101.39 \ (1)^{\circ}$	T = 20 (1) K
V = 571.8 (6) Å ³	Needle, colourless
Z = 2	0.4 \times 0.3 \times 0.2 mm
Data collection	
Huber diffractometer with Bruker	$R_{\rm int} = 0.027$
APEX CCD area detector	$\theta_{\rm max} = 30.0^{\circ}$
φ scans	$h = -13 \rightarrow 14$
13 577 measured reflections	$k = -5 \rightarrow 6$
1851 independent reflections	$l = -16 \rightarrow 16$
1811 reflections with $I > 2\sigma(I)$	
Refinement	
Refinement on F^2	$w = 1/[\sigma^2(F_0^2) + (0.0429P)^2]$
$R[F^2 > 2\sigma(F^2)] = 0.027$	+ 0.1445P]
$wR(F^2) = 0.071$	where $P = (F_0^2 + 2F_c^2)/3$
Huber diffractometer with Bruker APEX CCD area detector φ scans 13 577 measured reflections 1851 independent reflections 1811 reflections with $I > 2\sigma(I)$ <i>Refinement</i> Refinement on F^2 $R[F^2 > 2\sigma(F^2)] = 0.027$ $wR(F^2) = 0.071$	$R_{int} = 0.027$ $\theta_{max} = 30.0^{\circ}$ $h = -13 \rightarrow 14$ $k = -5 \rightarrow 6$ $l = -16 \rightarrow 16$ $w = 1/[\sigma^{2}(F_{o}^{2}) + (0.0429P)^{2} + 0.1445P]$ where $P = (F_{o}^{2} + 2F_{c}^{2})/3$

 $(\Delta/\sigma)_{\rm max} = 0.001$

 $\Delta \rho_{\rm max} = 0.37 \text{ e} \text{ } \text{\AA}^{-3}$

 $\Delta \rho_{\rm min} = -0.16 \text{ e} \text{ Å}^{-3}$

Table 1

Comparison of selected torsion angles (°) for L-alanylglycyl-L-alanine monohydrate, (I), and the water-free form, (II) (Padiyar & Seshadri, 1996).

For definitions of nomenclature, see IUPAC-IUB Commission on Biochemical Nomenclature (1970).

Torsion angle	Nomenclature	(I)	(II)
N1-C1-C2-N2	ψ_1	-146.8(2)	172.6 (2)
C1-C2-N2-C3	ω_1	-173.5(2)	-178.2(2)
C2-N2-C3-C4	φ_2	86.4 (2)	91.7 (1)
N2-C3-C4-N3	ψ_2	-167.4(2)	-151.9(2)
C3-C4-N3-C5	ω_2	-173.8(2)	-176.9(1)
C4-N3-C5-C6	φ_3	-159.1(2)	-71.3(2)
N3-C5-C6-O4	$\psi_{3.2}$	-5.0 (3)	-6.9 (1)

Table 2

Hydrogen-bond geometry (Å, °).

$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$					
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	$D - H \cdots A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdots A$
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	$N1 - H11A \cdots O4^{i}$ $N1 - H11B \cdots O3^{ii}$ $N1 - H11B \cdots O3^{ii}$	0.89 (2) 0.90 (2)	2.03 (2) 1.87 (2)	2.859 (2) 2.763 (2)	154 (2) 169 (2)
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	$N1 - H11B \cdots O4^{n}$ $N1 - H11C \cdots O4^{iii}$ $N2 - H12 \cdots O5^{iv}$	0.90(2) 0.93(2) 0.84(2)	2.59(2) 1.88(2) 2.02(2)	3.272 (2) 2.760 (2) 2.815 (2)	133 (2) 156 (2) 157 (2)
	$N3-H13\cdots O1^{i}$ $O5-H25A\cdots O2$ $O5-H25B\cdots O2^{iv}$	0.88 (2) 0.85 (2) 0.86 (2)	2.04 (2) 1.92 (2) 1.94 (2)	2.885 (2) 2.750 (2) 2.798 (2)	163 (2) 165 (2) 171 (2)

Symmetry codes: (i) -x + 1, $y + \frac{1}{2}$, -z + 2; (ii) x, y, z - 1; (iii) -x + 1, $y - \frac{1}{2}$, -z + 2; (iv) -x + 2, $y + \frac{1}{2}$, -z + 2.

All H atoms were found in difference Fourier maps and refined freely.

Data collection: *SMART* (Siemens, 1996); cell refinement: *SMART*; data reduction: *SAINT* (Siemens, 1996) and *SADABS* (Siemens, 1996); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEPIII* (Burnett & Johnson, 1996) and *SCHAKAL99* (Keller & Pierrard, 1999); software used to prepare material for publication: *PLATON* (Spek, 2003).

Supplementary data for this paper are available from the IUCr electronic archives (Reference: FA1135). Services for accessing these data are described at the back of the journal.

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All H-atom parameters refined

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